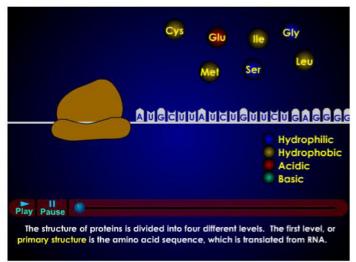
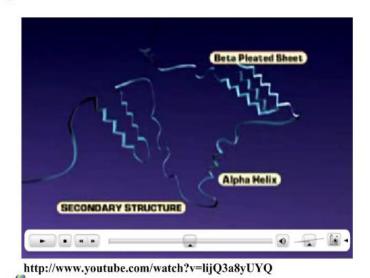


# Proteins (7.5 & C.1)

#### Protein folding and structure animations:



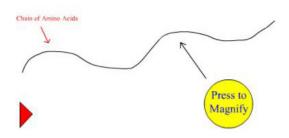
https://mywebspace.wisc.edu/jonovic/web/proteins/Proteins.swf



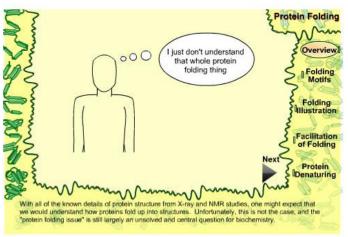
http://www.youtube.com/watch?v=lijQ3a8yUYQ

- http://www.stolaf.edu/people/giannini/flashanimat/proteins/protein%20structure.swf
- http://www.wiley.com//legacy/college/boyer/0470003790/
- http://www.stolaf.edu/people/giannini/flashanimat/proteins/hydrophobic%20force.swf

#### **Primary Structure**



http://www.stolaf.edu/people/giannini/flashanimat /proteins/protein%20structure.swf



http://www.wiley.com/legacy/college/boyer/0470003790/animations/protein\_folding/protein\_folding.swf

#### Proteins are the stable, folded 3D structure of polypeptides.

Polypeptides are the chain of amino acids that fold into proteins.

There are four levels of protein structure.

#### Primary (1°)

- amino acid sequence
- peptide bonds



#### Secondary (2°)

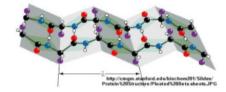
- repeating local structures
- held by H-bonds

#### α-helix



http://www.brooklyn.cuny.edu/bc/ahp/LAD /C4b/C4b\_proteinShape.html

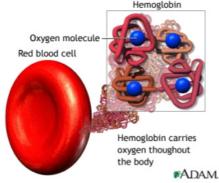
#### β-pleated sheet



#### Tertiary (3°)

- folding of a single protein
- R-group interactions
- hydrophobic core



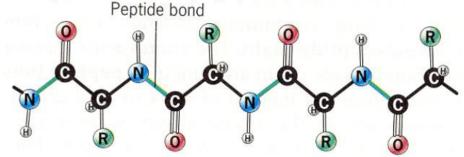


#### Quaternary (4°)

- protein complex
- made of two or more subunits

## Primary (1°) Structure of proteins: The polypeptide

How many amino acids in this polypeptide?

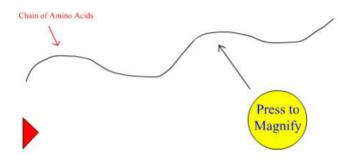


- Amino acid sequence
- Peptide bonds



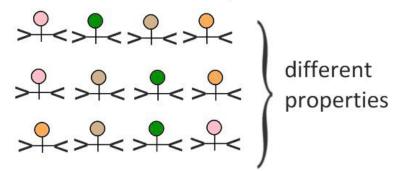
http://www.bio.miami.edu/~cmallery/255/255amino/255aminoacids.htm

### **Primary Structure**

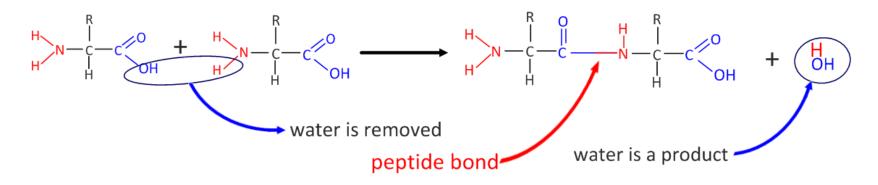


http://www.stolaf.edu/people/giannini/flashanimat /proteins/protein%20structure.swf Why are there infinite possibilities of polypeptides?

- 20 different amino acids
- could be any length
- amino acids can be in any order



#### Polypeptides are chains of amino acids joined by peptide bonds:



	H	H	H	H	H
	0,0	0, 1	0, ا	ا مر ا	0,0
	H <sub>3</sub> N <sup>+</sup> - °C - C @	H <sub>3</sub> N* - C - C G	H₃N+ - °C - C.⊕	H <sub>2</sub> N+ - °C - C @	H <sub>2</sub> N* - °C - C, O
	0 0	0	0 0	0 0	0
	(CH <sub>2</sub> ) <sub>3</sub>	ĊH <sub>2</sub>	CH <sub>2</sub>	CH <sub>2</sub>	CH.
	1 22	1	1 7	1 1	
	ŃН	ĊH <sub>2</sub>			
	i i	1			N.
	C=NH,	c=0	V		H
	U 11112	0-0		он	
	NH <sub>2</sub>	MH	Phenylalanine	Tyrosine	Tryptophan
		NH <sub>2</sub>			
20	Arginine	Glutamine	(Phe / F)	(Tyr / Y)	(Trp, W)
ī	(Arg/R)	(Gln / Q)	н	н	н
7			n .	n o	n o
<u>,                                    </u>	H	н			
õ		н	H₃N* -°C - C⊕	H₃N⁺ - °C - C⊕	H₃N* - °C - C⊕
·5	H <sub>3</sub> N* - °C - C⊗		0,	0	0
જે જ	0,0	H₃N* - °C - C.⊖	CH <sub>3</sub>	CH <sub>2</sub>	CH <sub>2</sub>
ے ہ' <u>ا</u>	(CH <sub>2</sub> ) <sub>4</sub>	0		HN N	
Ĕ.Ě		H			OH
<u> </u>	NH <sub>2</sub>	Glycine	Alanine	Histidine	Serine
5 7	Lysine	(Gly / G)	(Ala / A)	(His / H)	(Ser / S)
ર દે	(Lys/K)	H	H	Н	H
. <u>e</u> 6.	H <sub>2</sub>	0. 1	0. 1	0, 1	0. [
<u> </u>	127	H <sub>2</sub> N* - *C - C @	H <sub>2</sub> N* -*C - C @	H <sub>2</sub> N* -*C - C @	H <sub>2</sub> N* - *C - C 9
<u> </u>		1 0	1 0	1 0	1 0
∑ 200	H <sub>2</sub> C CH <sub>2</sub>	CH <sub>2</sub>	CH,	H-C-OH	CH <sub>2</sub>
<u> </u>	\ / \	1 1	1 1		1 1
ξ÷.	H <sub>2</sub> N* -*C - C ⊕	CH <sub>2</sub>	соон	CH <sub>2</sub>	SH
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edia.org/wikipedia/commons/ _acids_2.png/483px-Amino_acids_2.png	H	Glutamic Acid	Aspartic Acid	Threonine	Cysteine
ă <u>"</u> ı	0. 1	(Glu/E)	(Asp / D)	(Thr / T)	(Cys / C)
nttp://upload.wikimedia.org/wikipedia/commons :humb/c/c5/Amino_acids_2.png/483px-Amino_a	H <sub>2</sub> N* - °C - C @	Н	Н	Н	Н
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5 5	ĊH <sub>2</sub>	H,N* - *C - C.9	H,N* -*C - C.	H <sub>2</sub> N* -*C - C 9	H <sub>2</sub> N* - *C - C @
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も 5	i i	CH <sub>2</sub>	CH <sub>2</sub>	HC-CH <sub>3</sub>	
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:: <b>1</b>	ĭ	СН	C=0	CH <sub>2</sub>	
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트로			NH <sub>2</sub>	CH <sub>3</sub>	
	Methionine	Leucine	Asparagine	Isoleucine	Valine
	(Met/M)	(Leu/L)	(Am / N)	(Ile / I)	(Val / V)

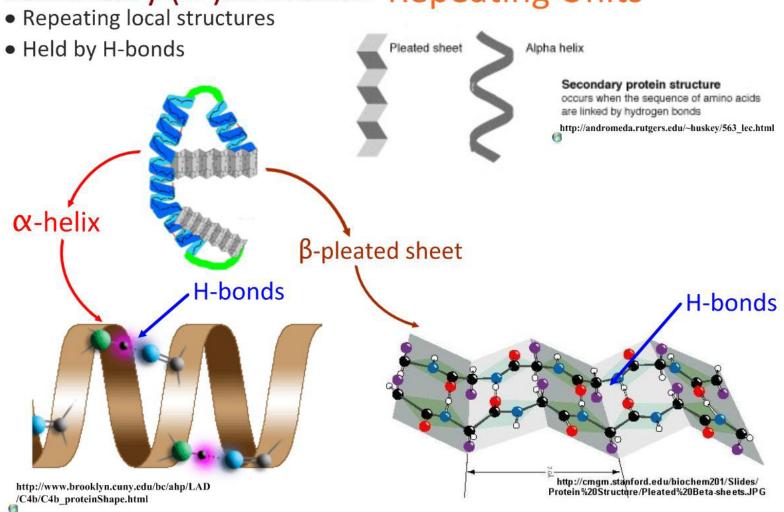
#### There are 20 different amino acids.

These can be combined in any order. Each amino acid has unique properties:

- some a polar (hydrophilic)
- some are non-polar (hydrophobic)
- some are positively or negatively charged
- some contain sulphur

The properties of the amino acids determine how a polypeptide folds up into a protein.

## Secondary (2°)structure: Repeating Units



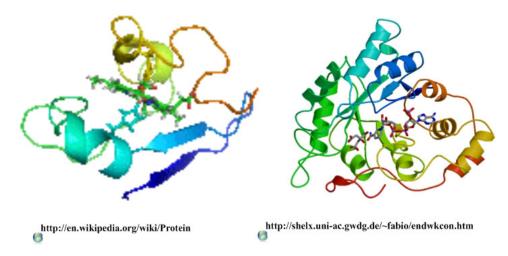
## Tertiary structure (3°): 3D Folded Structure

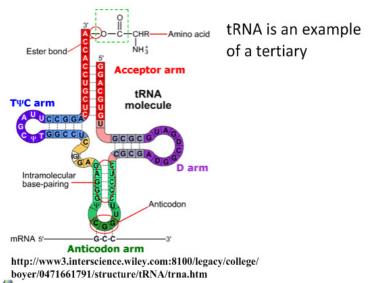
- Folding of a single protein into a 3D structure
- Hydrophobic core

This 3D structure gives proteins their functional properties, such as active sites on enzymes.



- -hydrophilic polar amino acids orient to the outside
- hydrophobic non-polar amino acids protected in the core
- oppositely-charged ions attract
- disulphide bridges formed between sulphur-containing amino acids





## Modeling protein folding

colour					
number	11	1	1	2	2
e.g.	serine	lysine	aspartic acid	cysteine	valine
	SER	LYS	ASP	CYS	VAL
property	polar	+ ve	- ve	contains S	non-polar

What is the primary structure of this section of a polypeptide?



## Modeling protein folding

try it with plasticene

colour					
number	11	1	1	2	2
e.g.	serine	lysine	aspartic acid	cysteine	valine
	SER	LYS	ASP	CYS	VAL
property	polar	+ ve	- ve	contains S	non-polar

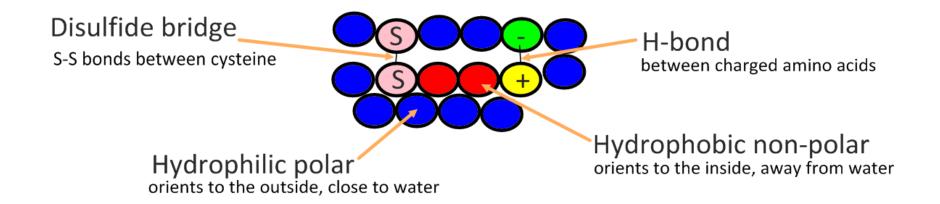
What is the primary structure of this section of a polypeptide?



How would the polypeptide fold if placed in water?

- 1. polar groups outside
- 2. non-polar groups inside
- 3. positive near negative
- 4. disulfide bridges

#### Modeling protein folding:

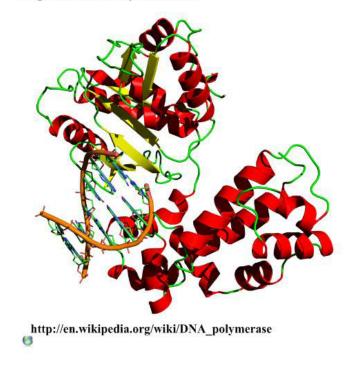


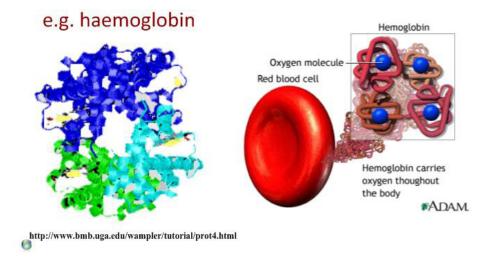
colour					
number	11	1	1	2	2
e.g.	serine	lysine	aspartic acid	cysteine	valine
	SER	LYS	ASP	CYS	VAL
property	polar	+ ve	- ve	contains S	non-polar

## Quaternary structure (4°): Multiple Subunits

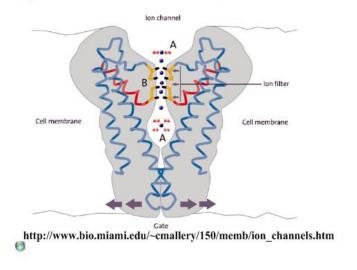
- Protein complex
- Made of two or more subunits

#### e.g. DNA Polymerase



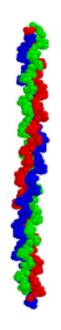


#### e.g. Ion channels in the plasma membrane



 The heme is a small but important non-protein molecule, or prosthetic *group*, that binds an *iron* atom in the ferrous (+2) oxidation state.

## Fibrous Proteins



insoluble in water

structural (support, strength)

keratin (hair/nails)

elastin (skin)

collagen (tissue strengthening)

### **Globular Proteins**

can be soluble

functional (enzymes, antibodies)

amylase (digests starch)



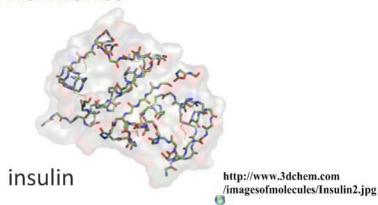
haemoglobin

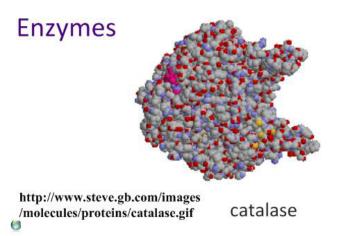
immunoglobulins (antibodies)

transport (Na<sup>+</sup>/K<sup>+</sup> pump)

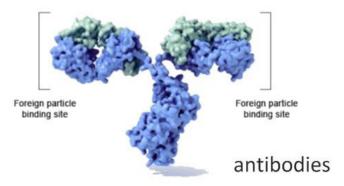
Uses of proteins:
Proteins can be structural or part of the plasma membrane. They can also perform other diverse functions in an organism:

#### Hormones

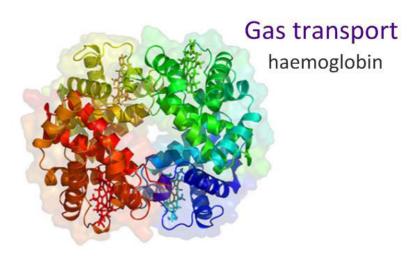




#### Immunoglobulin Immunoglobulin G (IgG)

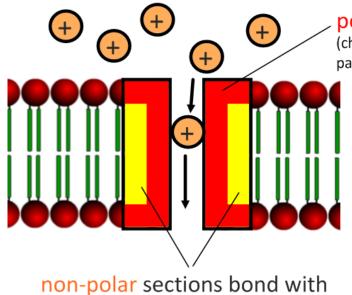


http://ghr.nlm.nih.gov/handbook/illustrations/igg.jpg



http://chemistry.ewu.edu/jcorkill/biochem/48006.htm

#### Polar and non-polar amino acids are used in membrane channels:

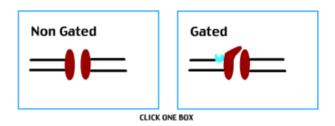


hydrophobic tails

polar sections allow ions through (charged or polar molecules cannot normally pass through the hydrophobic layers)

polar amino acids are positioned on external surfaces and line the protein channels for facilitated diffusion

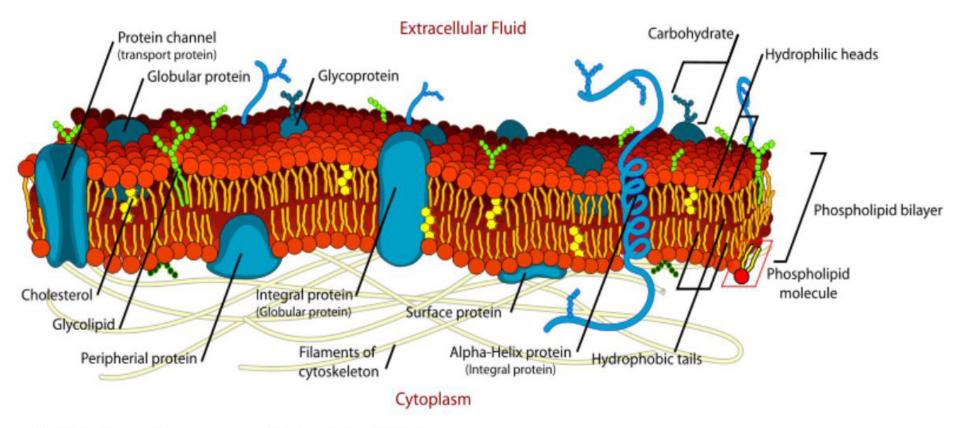
### **CHANNEL**



http://www.stolaf.edu/people/giannini/flashanimat/transport/channel.swf

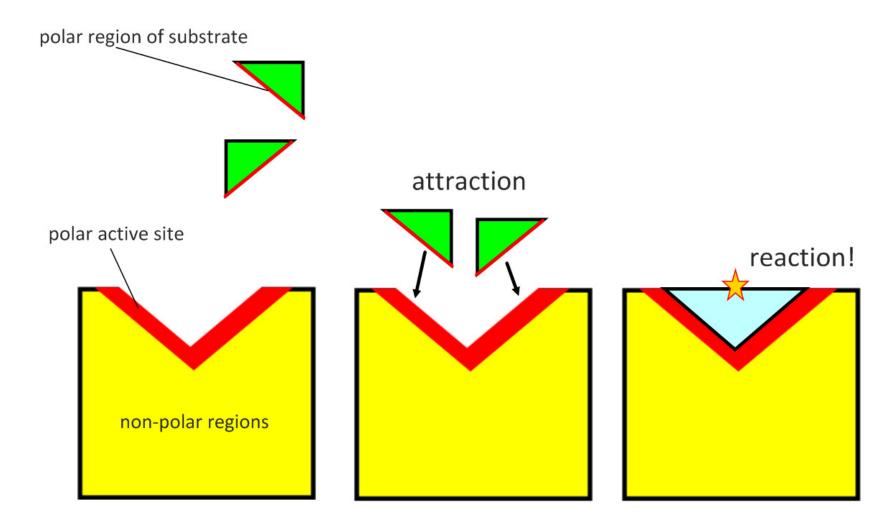
 http://www.stolaf.edu/people/giannini/flash animat/transport/channel.swf

## Non-polar amino acids help bond proteins to the plasma membrane

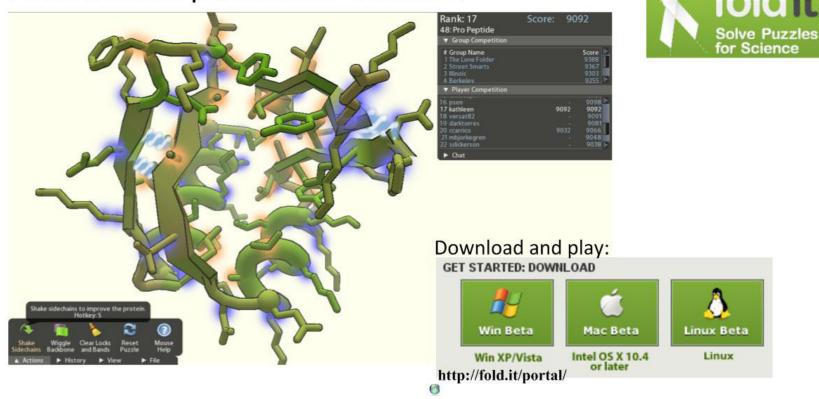


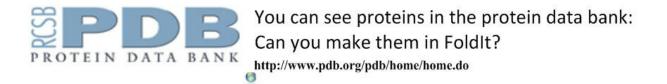
http://cellbiology.med.unsw.edu.au/units/science/lecture0803.htm

#### Polar amino acids in enzymes help bind the substrates for reaction:



## Foldit: Solve puzzles for Science!





http://fold.it/portal/